MASTER[®] SAFETY DATA SHEET

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1. Product Identification

Master Products 4635 Willow Drive Medina, MN 55340 (612) 478-2360

Product line:
Products:
CAS:
Synonyms:
Recommended use:
Restrictions:
Created:
Revised:
Emergency phone:

MASTER ® DOT 3 Brake Fluid

FH12, FH32, FH128
Not applicable (Mixture)
Glycol-Based Brake Fluid
Disk and drum hydraulic brake fluid
Do not use where DOT5 is specified
6 April 2012
26 March 2015
CHEMTREC: (+1) 800-424-9300

2. Hazards Identification

Appearance:	Clear, pale yellow liquid
Odor:	Mild, sweet odor
Classification(s):	Acute Toxicity, Oral Category 4*
	Skin Irritation, Category 2
	Eye Irritation, Category 2A
	Target Organ Toxicity, Acute Category 2
Target organs:	Kidney, Liver, Central Nervous System

Target organs Symbol(s):



Signal Word: Hazard Statement(s):	Warning Harmful if swallowed. Causes mild skin irritation. Causes serious eye irritation.
Other hazard(s):	Combustible liquid. Repeated exposure may cause dryness of the skin. Vapors may cause respiratory irritation.

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Precaution(s):	Wear eye and skin protection before handling. Do not breathe mist/vapors/spray. Use in a well ventilated area. Wear protective gloves/protective clothing. IF IN EYES:
Disposal:	Flush with water for 15 minutes and consult a physician. Do no ingest. IF SWALLOWED: Do NOT induce vomiting. Immediately call a POISON CENTER or doctor/physician. Keep out of waterways. Check local, national, and international regulations for proper disposal

HMIS (estimated): Health – 3 Fire – 1 Instability – 0

*Classified based on human experience and epistemological data, not based on strict application of the GHS criteria

3. Composition/Information on Ingredients

Hazardous Ingredients:

Component	CAS No.	Conc (wt%)
Diethylene Glycol	111-46-6	20 - 40
2-(2-propoxyethoxy)ethanol	6881-94-3	0 – 30
2-(2-butoxyethoxy)ethanol	112-34-5	0 – 20
Ethoxytriglycol	112-50-5	0 – 20
Butoxytriglycol	143-22-6	30 – 70
Additives	Proprietary	< 1

4. First Aid Measures		
Eyes	Remove contact lenses, if worn. Rinse with running water for at least 15 minutes, lifting upper and lower eyelids occasionally. Seek medical attention.	
Skin	Remove affected clothing and launder before reuse. Wash affected area for at least 15 minutes with soap and running water. Prolonged or repeated exposure may cause defatting of the skin – symptoms include redness, dryness, cracking	
Inhalation	Remove exposed person to fresh air immediately. Restore or assist breathing, if necessary. Get medical attention immediately if symptoms of CNS depression or intoxication develop	
Ingestion	Do NOT induce vomiting. If conscious, give two full glasses of water. If a significant volume has been swallowed, get medical attention immediately.	

	Swallowing large amounts of diethylene glycol is potentially lethal. Immediate symptoms may include severe abdominal cramping, diarrhea, vomiting, intoxication, and hypertension. Infrequent urination and other cardiac, neurological, and renal effects of metabolic acidosis, hyponatremia, or hyperkalemia may develop. Diethylene glycol has been known to cause metabolic acidosis leading to kidney and liver failure, neurological complications, and death.
Additional Info	Note to physician: Treat for diethylene glycol poisoning
Specific Treatments	Immediately treat with hemodialysis. Diethylene glycol is metabolized by NAD-dependent alcohol dehydrogenase and aldehyde dehydrogenase into 2-hydroxyethoxyacetadlehyde and 2-hydroxyethoxyacetic acid, respectively. Administering NAD-dependent alcohol dehydrogenase inhibitors such as ethanol or fomepizole may slow the production of harmful metabolites.

5. Fire Fighting Measures

NFPA (estimated):	Health – 2 F	Fire – 1	Instability – 0
Flash Point	93°C / 199°F (calculate	ed)
Extinguishing Media	large fires app	ly large (phol foam, dry chemical or CO ₂ . For (flooding) quantities of water from as a spray or mist.
Unsuitable Media	Water jet may	be ineffe	ctive
Firefighting Procedures: Wear a self-container breathing apparatus if necessary based on concentrations of smoke. Material will produce primarily oxides of carbon as combustion products.			
Unusual Hazards	Not Determine	ed	

6. Accidental Release Measures

Personal precautions, protective equipment, and emergency procedures: Ventilate if released in a confined area. Avoid breathing mists/vapors/spray. Product may present slipping hazard if left on the floor. Beware of vapors pooling in low areas to explosive concentrations. **Environmental precautions:** Avoid release to the environment. Prevent from entering into soil, ditches, sewers, waterways or groundwater

Methods for removal: Use an explosion-proof pump to remove bulk liquid. Residual liquid can be absorbed on inert material. Dispose of contaminated adsorbent as hazardous waste. Wash the area with water after excess product and adsorbent is removed.

7. Handling and Storage

Max. Handling Temp: Not determined

- Procedures: Use in a well ventilated area. Avoid breathing mists/vapors/spray. Avoid handling hot product where possible. Use appropriate personal protective equipment to avoid contact with skin and eyes. Note the location of nearest emergency shower and eye wash station before use. Store with the lid tightly closed in a cool, dry, well-ventilated place. Product is hygroscopic and effectiveness may diminish if opened product is stored for long periods of time. Dispose of spilled or used material in accordance with local, regional, national, and international regulations.
- **Max Store Temp:** Do not store or handle at elevated temperatures.

8. Exposure Controls/Personal Protection

Exposure Limits

US

Guidelines by component Diethylene Glycol (CAS# 111-46-6) OSHA TWA: 10mg/m3 Ethanolamine (CAS# 141-43-5) ACGIH TWA: 3 ppm ACGIH STEL: 6 ppm OSHA TWA: 3 ppm OSHA STEL: 6 ppm NIOSH TWA: 3 ppm

Other Exposure Limits: Not determined

Engineering Controls: Use in a well ventilated area. Local and general ventilation should keep methanol vapor concentration below permissible limits. Where exposure potential exceeds recommended limits, use a NIOSH/OSHA approved supplied air respirator

as recommended. Vapors are heavier than air and will tend to accumulate in low-lying areas.

Personal Protective Equipment

- **Respiratory:** Use a NIOSH or CEN approved full-face respirator with multipurpose combination or type ABEK respirator cartridges as a backup to engineering controls. If the respiratory is the only means of protection, use a full-face supplied air respirator
- **Eye:** Use tightly-fitting chemical splash goggles. Use face shield, especially where splashing is likely to occur
- **Gloves:** Use nitrile, butyl, viton, or fluoroelastemer gloves. Even appropriate materials may degrade after prolonged exposure with product.
- Clothing: Use chemical resistant pants and jackets, preferably of butyl or nitrile rubber
- Other: Locate the nearest eyewash station and safety shower before handling this product. Limit exposure whenever possible.

Hygiene: Wash thoroughly after handling this product.

9. Physical and Chemical Properties

Appearance Odor Odor threshold pH Melting Point Initial Boiling Pt	Clear, pale yellow liquid Mild, sweet odor Not determined 7 - 11 < -50°C / -58°F > 210°C / 410°F
Flash Point	93°C / 199°F
Evaporation Rate	
Upper Flammable Lm	
Lower Flammable Lm	Not determined
Explosive Data	Vapors may form explosive mixtures with air
Vapor Pressure	0.09 hPa (0.07 mmHg) @ 20° (68°F)
Vapor Density	> 5 (Air = 1)
Volatile Organics	Not determined
Density	1.05 mg/cu. cm @15.6°C
Solubility	Miscible in water, alcohol; sparingly soluble in some organic solvents
K _{ow} Viscosity	Not determined 1.8 mm/s ² @ 100°C
Autoignition Point Decomposition Temp	Not determined

10. Stability and Reactivity

Stability	Material is normally stable at ambient temperatures and pressures.	
Decomposition Temp Incompatibility	•	
Polymerization	Will not occur	
Thermal Decomposition Primarily oxidizes to carbon dioxide in normal combustion		
Conditions to Avoid	conditions. In lower oxygen environments carbon monoxide, formaldehyde, or formic acid may be formed. Vapors may catch fire – keep away from strong oxidizers, acids, bases as well as heat/sparks/open flames/hot surfaces	

11. Toxicological Information

- Acute Exposure –		
Eye Irritation	Expected to cause mild to moderate irritation of the eye if exposed to liquid or in high vapor concentrations. May cause	
	irritation, tearing, or burning of the eyes.	
Skin Irritation	Expected to be mildly irritating to the skin. Symptoms of irritation may include redness, drying, and cracking of the	
	skin.	
Respiratory Irritation	High vapor concentrations may cause transient irritation to the respiratory system.	
Dermal Toxicity	This product can be absorbed through the skin, but is of low order of toxicity. Limit exposure to skin where possible.	
Inhalation Toxicity	Toxicity is similar to that for oral ingestion, though this exposure mode is far less likely to occur.	
Oral Toxicity	Toxic or fatal if ingested. Symptoms of diethylene glycol poisoning include severe abdominal cramping, diarrhea, vomiting, sweating, confusion, cardiac abnormalities, neurological abnormalities, infrequent urination, intoxication or CNS depression. If left untreated, product will metabolize to cause metabolic acidosis, renal failure, hyperkalemia, hyponatremia, parylsis, cardiac failure, or death. Seek medical attention immediately for poisoning. If ingested, DO NOT wait for symptoms to develop before getting treatment.	
Aspiration Hazard	This product has a very low viscosity and may be fatal if aspirated into the airways. Do NOT induce vomiting, as this increases risk of aspiration.	
- Chronic Exposure –		
Chronic Toxicity	This product may cause dryness or defatting of the skin	

Chronic Toxicity This product may cause dryness or defatting of the skin, dermatitis, or may aggravate existing skin conditions.

Carcinogenicity	This product and its components are NOT listed by the IARC, NTP, ACGIH, or OSHA as carcinogens
Mutagenicity	Available information does not suggest that this product is a germ cell mutagen
Reproductive Toxicity	Available information does not suggest that this product is a reproductive toxin.
Teratogenicity	Diethylene glycol has produced birth defects in rats at concentrations that are toxic to the mother.
	- Additional Information –
Target organ toxicity	Product is toxic to organs: Kidneys, liver, central nervous system, heart. Metabolic products of diethylene glycol produce acidosis and organ toxicity effects. In some cases, other metabolic abnormalities have been reported such as hyponatremia and hyperkalemia leading to nerve and cardiac damage.
Synergistic effects	Though specific data is not available, ethanol is a competing substrate for NAD-dependent alcohol dehydrogenase and may slow the product of harmful metabolic products of diethylene glycol.
Pharmacokinetics	No data available

12. Ecological Information

- Environmental Toxicity –

te LD50 > 75.2 g/L (96h)					
Freshwater Invertebrates Acute LD50 > 10g/l (24h)					
Not determined					
Not determined					
Saltwater Invertebrates Not determined					
Not determined					
determined					

- Environmental Fate –BiodegradationNo data available. Expected to biodegrade rapidly and
degrade by photo-oxidative reactions with the airBioaccumulationProduct is very mobile in soil and water and is somewhat
volatile – it is not expected to bioaccumulate.Soil MobilityProduct has high mobility in soil, slowly evaporates at
environmentally relevant temperaturesOther EffectsNot determined

13. Disposal Considerations

Disposal Considerations

All disposal practices must be in accordance with local, regional, national, and international regulations. Store material for disposal as indicated in Section 7.

Disposal by controlled incineration or by secure land fill may be acceptable – review applicable regulations or regulatory bodies before making disposal decisions.

Contaminated Containers or Packaging

Empty containers are likely to contain flammable vapors or explosive mixtures of vapor and air. Do NOT weld, cut, or grind empty containers. Rinse empty containers with water and dispose of in accordance with local, regional, national, and international regulations

14. Transportation Information

Description shown may not apply to all shipping situations. Consult applicable shipping codes to determine any additional shipping requirements

- Global Chemical Inventories/Regulations –				
15. Regulatory Information				
ICAO/IATA	Not dangerous goods			
IMDG	Not dangerous goods			
US DOT	Not dangerous goods			

- Global Chemical Inventories/Regulations –					
USA	All components of	this material are on the US TSCA			
Other TSCA Reg.	None known				
EU	registered under F Agency regarding legal requirements the EU.	is product and similar mixtures are REACH. Consult the European Chemicals REACH registration, reporting, and other s for methanol solutions before importing to			
New Zealand	May require notific Regulations	cation before sale under New Zealand			
Canada	All components of Domestic Substar	this product are listed on the Canadian nces List (DSL).			
Canada WHMIS	B3				
	- Other U.S. Fed	eral Regulations –			
SARA Ext. Haz. Subst	. No components listed as Extremely Hazardous Substances list. See 40 CFR 355				
SARA Sect. 313	ethoxytriglycol (C/	ethanol (CAS # 112-34-5) and AS # 112-50-5) are subject to reporting III, Section 313. See 40 CFR 372			
SARA 311/312 Class	Acute Hazard	- YES			

	Reactivity Hazard - NO		
CERCLA Haz. Sub.	No components listed.	See 40 CFR 302	

- State Regulations – CA Prop 65 This product does not contain any chemicals known to the State of California to cause cancer, birth defects, or any other reproductive harm.

Right to Know Component	Right to Know States
2-(2-propoxyethoxy)ethanol (CAS # 6881-94-3)	NJ, PA
Diethylene glycol (CAS # 111-46-6)	NJ, PA
Butoxytrigycol (CAS # 134-22-6)	NJ, PA
Ethoxytriglycol (CAS # 112-50-5)	NJ, PA
Poly(1,2-dihydro-2,2,4-trimethylquinoline) (CAS # 26780-96-1)	NJ, PA
2-(2-butoxyethoxy)ethanol (CAS # 112-34-5)	NJ, PA
Ethanolamine (CAS # 141-43-5)	NJ, PA, MA
Benzotriazole (CAS # 95-14-7)	NJ, PA, MA
Sodium Nitrate (CAS # 7631-99-4)	NJ, PA

- Other –

16. Other Information

Revision updates may be in many sections and the MSDS should be read in its entirety. Prepared according to the UN Globally Harmonized System for the Classification and Labeling of Chemicals (GHS).

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