

SAFETY DATA SHEET

1. SUBSTANCE AND SOURCE IDENTIFICATION

Product Identifier

SRM Number: 3177

SRM Name: Mercuric Chloride (HgCl₂) Standard Solution

Other Means of Identification: Not applicable.

Recommended Use of This Material and Restrictions of Use

This Standard Reference Material (SRM) is intended for use as a primary calibration standard for the quantitative determination of mercury when the chemical form of mercury is high-purity mercuric chloride (mercury (II) chloride). A unit of SRM 3177 consists of five 10 mL sealed borosilicate glass ampoules of an acidified aqueous solution prepared gravimetrically from high-purity mercury (II) chloride to contain a known mass fraction of mercury. The solution contains nitric acid at a volume fraction of approximately 3 % and hydrochloric acid at a volume fraction of approximately 4 %.

Company Information

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2. HAZARDS IDENTIFICATION

Classification

Physical Hazard: Not classified.

Health Hazard: Acute Toxicity, Oral Category 4
Skin Corrosion/Irritation Category 1B

Serious Eye Damage/Eye Irritation Category 1

Label Elements Symbol



Signal Word DANGER

Hazard Statement(s)

H302 Harmful if swallowed.

H314 Causes severe skin burns and eye damage.

Precautionary Statement(s)

P260 Do not breathe fumes, mists, vapors, or spray. P264 Wash hands thoroughly after handling.

P270 Do not eat, drink or smoke when using this product.

P280 Wear protective gloves, protective clothing, and eye protection.

P301+P330+P331 If swallowed: Rinse mouth. Do NOT induce vomiting.

P303+P361+P353 If on skin (or hair): Remove immediately all contaminated clothing. Rinse skin with

water.

P304+P340 If inhaled: Remove person to fresh air and keep comfortable for breathing.

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P305+P351+P338	If in eyes: Rinse cautiously with water for several minutes. Remove contact lenses,
	if present and easy to do. Continue rinsing.
P310	Immediately call a doctor.
P363	Wash contaminated clothing before reuse.
P405	Store locked up.
P501	Dispose of contents and container according to local regulations

Hazards Not Otherwise Classified: Not applicable.

Ingredients(s) with Unknown Acute Toxicity: Not applicable.

3. COMPOSITION AND INFORMATION ON HAZARDOUS INGREDIENTS

Substance: Nitric Acid/Hydrochloric Acid/Mercury (II) Chloride Solution

Other Designations:

Nitric Acid (aqua fortis; hydrogen nitrate; azotic acid; engraver's acid)

Hydrochloric Acid (hydrogen chloride acid)

Mercury (II) Chloride [mercury (II) dichloride; mercury (II) (2+) salt (2:1); mercuric chloride, mercury chloromercurate (II); corrosive sublimate; bichloride of mercury; corrosive mercury chloride]

NOTE: Mercury (II) chloride in hydrochloric acid and nitric acid forms a solvated mercury (II) chloride salt. The health and physical hazard information provided in this MSDS are for hydrochloric acid, nitric acid, and mercury (II) chloride.

Components are listed in compliance with OSHA's 29 CFR 1910.1200; for the actual values see the NIST Certificate of Analysis.

Hazardous Component(s)	CAS Number	EC Number (EINECS)	Nominal Mass Concentration (%)
Nitric acid	7697-37-2	231-714-2	3
Hydrochloric acid	7647-01-0	231-595-7	1.7
Mercuric (II) chloride	7487-94-7	231-299-8	0.1
Non-Hazardous Component(s) Water	7732-18-5	231-791-2	>95

4. FIRST AID MEASURES

Description of First Aid Measures:

Inhalation: If adverse effects occur, remove to uncontaminated area. Seek immediate medical attention.

Skin Contact: Wash skin with soap and water for at least 15 minutes while removing contaminated clothing and shoes. Seek immediate medical attention. Destroy contaminated shoes.

Eye Contact: Immediately flush eyes, including under the eyelids with copious amounts of water for at least 15 minutes. Seek immediate medical attention.

Ingestion: Contact a poison control center immediately for instructions. Give water to rinse out mouth. Never give liquids to a person with reduced awareness or becoming unconscious. If vomiting occurs, keep head lower than hips to prevent aspiration. If not breathing, give artificial respiration by qualified personnel. Seek immediate medical attention.

Most Important Symptoms/Effects, Acute and Delayed: Acid burns to skin, eyes, and lungs. Mercuric chloride is highly toxic (concentration in this SRM is 0.1 %).

Indication of any immediate medical attention and special treatment needed, if necessary: If any of the above symptoms are present, seek immediate medical attention.

5. FIRE FIGHTING MEASURES

Fire and Explosion Hazards: Negligible fire hazard. See Section 9, "Physical and Chemical Properties" for flammability properties.

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Extinguishing Media:

Suitable: Use extinguishing media appropriate to the surrounding fire.

Unsuitable: None listed.

Specific Hazards Arising from the Chemical: Thermal decomposition products: hydrogen chloride, acid halides.

Special Protective Equipment and Precautions for Fire-Fighters: Avoid inhalation of material or combustion byproducts. Wear full protective clothing and NIOSH approved self-contained breathing apparatus (SCBA).

Health = 3 Fire = 0 Reactivity = 0

6. ACCIDENTAL RELEASE MEASURES

Personal Precautions, Protective Equipment and Emergency Procedures: Immediately contact emergency personnel. Keep unnecessary personnel away. Use suitable protective equipment; see Section 8, "Exposure Controls and Personal Protection".

Methods and Materials for Containment and Clean up: Do not touch spilled material. Notify safety personnel of spills. Absorb with sand or other non-combustible material. Collect spilled material in appropriate container for disposal. Isolate hazard area and deny entry.

7. HANDLING AND STORAGE

Safe Handling Precautions: Handle glass ampoules with care. See Section 8, "Exposure Controls and Personal Protection".

Storage: Store and handling in accordance with all current regulations and standards. Keep separated from incompatible substances (see Section 10, "Stability and Reactivity").

8. EXPOSURE CONTROLS AND PERSONAL PROTECTION

Exposure Limits				
Components	OSHA (PEL)	ACGIH (TLV)	NIOSH (REL)	
Nitric acid	TWA: 5 mg/m ³ (2 ppm)	TWA: 5 mg/m ³ (2 ppm) STEL: 10 mg/m ³ (4 ppm)	TWA: 5 mg/m ³ (2 ppm) STEL: 10 mg/m ³ (4 ppm) IDLH: 65 mg/m ³ (25 ppm)	
Hydrochloric Acid	Ceiling: 7 mg/m ³ (5 ppm)	Ceiling: 3 mg/m³ (2 ppm)	Ceiling: 7 mg/m ³ (5 ppm) IDLH: 50 ppm	
Mercury (II) Chloride (as total Hg)	TWA: 0.1 mg/m ³	TWA: 0.025 mg/m³ (related to Mercury inorganic compounds) Skin - potential significant contribution to overall exposure by the cutaneous route (related to Mercury inorganic forms)	TWA: 0.05 mg/m³ (vapor), 0.1 mg/m³ ceiling, IDLH: 10 mg/m³, except organo (alkyl) compounds as Hg, related to mercury compounds Potential for dermal absorption (related to Mercury compounds)	

Engineering Controls: Provide local exhaust or process enclosure ventilation system. Ensure compliance with applicable exposure limits.

Personal Protection: In accordance with OSHA 29 CFR 1910.132, subpart I, wear appropriate Personal Protective Equipment (PPE) to minimize exposure to this material.

Respiratory Protection: If workplace conditions warrant a respirator, a respiratory protection program that meets OSHA 29CFR 1910.134 must be followed. Refer to NIOSH 42 CFR 84 for applicable certified respirators.

Eye/Face Protection: Wear splash resistant safety goggles with a face shield. An eyewash station should be readily available near areas of use.

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Skin and Body Protection: Personal protective equipment for the body should be selected based on the task being performed and the risks involved and should be approved by a specialist before handling this product. Chemical-resistant gloves should be worn at all times when handling chemicals.

9. PHYSICAL AND CHEMICAL PROPERTIES

Descriptive Properties					
Appearance (physical state, color, etc.):	colorless to yellow liquid				
Molecular Formula	solution, not applicable				
Molar Mass (g/mol)	solution, not applicable				
Odor	irritating odor				
Odor threshold	not available				
рН	Acidic				
Evaporation rate (ether = 1)	not available				
Melting point/freezing point	not available				
Sublimation Point	not applicable				
Relative Density as specific gravity (water = 1)	~1				
Vapor Pressure	not available				
Vapor Density (air = 1)	not available				
Viscosity (cP)	not available				
Solubility(ies)	miscible with water and ethanol				
Partition coefficient (n-octanol/water)	not available				
Thermal Stability Properties:					
Autoignition Temperature	not applicable				
Thermal Decomposition	not applicable				
Initial boiling point and boiling range:	~100 °C (212 °F)				
Explosive Limits, LEL (Volume %)	not applicable				
Explosive Limits, UEL (Volume %)	not applicable				
Flash Point	not applicable				
Flammability (solid, gas)	not applicable				
10. STABILITY AND REACTIVITY					
Reactivity: Stable at normal temperatures and pressure.					
Stability: X Stable Unstable	2				
Possible Hazardous Reactions: May react with evolution gases on contact with water.	of heat; release toxic, corrosive, flammable or explosive				
Conditions to Avoid: Heat, flames, sparks and other sources of ignition. May ignite or explode on contact with combustible materials.					
Incompatible Materials: Cyanides, metals, amines, bases, metal carbide, oxidizing materials, acids, halo carbons, combustible materials, halogens, metal salts.					
Fire/Explosion Information: See Section 5, "Fire Fighting Measures".					
Hazardous Decomposition: Thermal decomposition will produce hydrogen chloride gas, chlorine.					
Hazardous Polymerization: Will Occur X Will Not Occur					

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11. TOXICOLOGICA	AL INI	FORMATION			
Route of Exposure:	X	_ Inhalation	X	Skin	Ingestion
					cal Characteristics: Burning pain, severe skin ially fatal on contact with the skin or swallowed.
Short term exposure throat, shortness of I to teeth, bronchial in Chloride: Short-ter vomiting, diarrhea, exposure may cause Skin Contact: Hydon the concentration Short-term and long Eye Contact: Hydon	chlorice may coreath, ritation expore chest is blue lidrochlon and correction cochloric rochloric rochlori	e acid and nitric a cause irritation and headache, dizzine n, chronic cough, osure may cause pain, difficulty b ines on gums, loo oric acid and nitric duration of expos exposure may cau ric acid and nitric	cid can of inflamings, and repronchia irritation reathing sening of a cid calure. Effects is irritated acid calure	damage the mation of the nausea. Lord l pneumoni (possibly son headache, of teeth, nerrous cause severects of acidion (possibut cause severects)	e mucous membranes and upper respiratory tract, he upper respiratory tract, coughing, choking, soreing term exposure to acid fumes may cause damage ia, and gastrointestinal disturbances. Mercury (II) severe), allergic reactions, metallic taste, nauseau, lung damage, and kidney damage. Long-terminated damage, and reproductive effects. Vere skin burns. Severity of the damage dependent of burns may be delayed. Mercury (II) Chlorides oly severe), allergic reactions, and death. Evere eye irritation, corneal burns, permanent eyes
Mercury (II) chlorid	le: Sho	ort-term and long-	term ex	posure may	on the concentration and duration of exposure. v cause irritation (possibly severe). e burns to the gastrointestinal tract. Mercury (II)
9					including nerve damage, tumors.
Numerical Measures of	f Toxic	eity:			
Hydrochloric a Hydrochloric a Mercuric (II) cl	, Inhala cid: Ra cid: Ra nloride	v 4, Oral ation LC50: 130 tt, Inhalation LC5 tt, Oral LD50: 70 : Rat, Oral LD50 : Rat, Dermal LD	0: 3124 0 mg/kg : 1 mg/k	ppm (1 h);	; 1562 ppm (4 h)
Skin Corrosion/Iri This SRM cont			acid and	nitric acid	and it is classified as Category 1B.
Serious Eye damag This SRM cont				nitric acid	and it is classified as Category 1.
Respiratory Sensit	ization	: No data availal	ole.		
Skin Sensitization:	No da	ata available.			
Germ Cell Mutage	nicity:	No data availab	le.		
Nitric acid is no Hydrochloric a NTP or OSHA Mercu	rcinogot listed and as a caric (II)	en/Potential Car d by NTP, IARC of Mercuric (II) chl arcinogen/potential chloride: Mutag	or OSHA oride are l carcine enic – B	A as a carcine listed by Logen. accillus sub	Yes X No inogen or a potential carcinogen. [ARC as group 3, not classifiable and not listed by btilis: 50 mmol/L ral (TDLo): 5150 mg/kg (103 wk)
Nitric acid, Rat Nitric acid, Rat	cid: Ra , Oral ' , Oral '	tt, Oral TCLo: 45 TDLo: 21 150 mg/ TDLo: 2345 mg/	g/kg (pre kg (preg	egnant 1 d t nant 18 d)	to copulation 1 d) to 21 d) m ³ (4 h, pregnant 9 d–12 d)
Specific Target Or	gan To	oxicity, Single Ex	posure:	Not classi	ified.
Specific Target Or	gan To	oxicity, Repeated	Exposi	ıre: Not cl	lassified.

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Aspiration Hazard: No data available.

12. ECOLOGICAL INFORMATION

Ecotoxicity Data

Nitric acid

Starfish (Asterias rubens) LC50 [renewal/aerated water]: 100 mg/L to 300 mg/L (48 h)

Hydrochloric acid

Fish Toxicity: Mosquitofish (*Gambusia affinis*) LC50 (static): 282 mg/L (96 h) Invertebrate: Shore crab (*Carcinus maenas*) LC50 (mortality): 240 mg/L (48 h)

Mercury (II) chloride

Fish Toxicity: Rainbow trout (Oncorhynchus mykiss) LC50 (static): 0.13 mg/L - 0.19 mg/L (96 h)

Persistence and Degradability: No data available.

Bioaccumulative Potential: No data available.

Mobility in Soil: Hydrogen chloride will evaporate from dry soil surfaces and dissociate into chloride and hydronium ions in moist soil.

ions in moist son.

Other Adverse effects: No data available.

13. DISPOSAL CONSIDERATIONS

Waste Disposal: Dispose of waste in accordance with all applicable federal, state, and local regulations. Hydrochloric acid subject to disposal regulations: U.S. EPA 40 CFR 262, Hazardous Waste Number: D002.

Dispose in accordance with all applicable federal, state, and local regulations. Mercury Hazardous Waste Number D009. Dispose of in accordance with U.S. EPA 40 CFR 262 for concentrations at or above the Regulatory level (0.2 mg/L).

14. TRANSPORTATION INFORMATION

U.S. DOT and IATA: UN1760, Corrosive liquid n.o.s. (contains nitric acid and hydrochloric acid), Hazard Class 8, Packing Group II.

15. REGULATORY INFORMATION

U.S. Regulations

CERCLA Sections 102a/103 (40 CFR 302.4): Nitric Acid, 1000 lbs (454 kg) final RQ.

Hydrochloric Acid, 5000 lb (2270 kg) final RQ.

SARA Title III Sections 302 (40 CFR 355.30): Nitric Acid, 1000 lbs TPQ.

Hydrochloric Acid, 500 lb TPQ (gas only).

Mercury (II) Chloride, 500 lbs lower TPQ and 10 000 lbs upper TPQ.

SARA Title III Sections 304 (40 CFR 355.40): Nitric Acid, 1000 lbs EPCRA RQ.

Hydrochloric Acid, 5000 lb RQ (gas only).

Mercury (II) Chloride, 500 lbs EPCRA RQ.

SARA Title III Sections 313 (40 CFR 372.65): Nitric Acid, 1.0 % de minimis concentrations.

Hydrochloric Acid, 1.0 % de minimis concentrations; (acid aerosols including mists, vapors, gas, fog, and other airborne forms of any particle size).

Mercury (II) Chloride, 1.0 % de minimis concentrations, supplier notification limit (Chemical category N458, related to Mercury compounds).

OSHA Process Safety (29 CFR 1910.119): Nitric Acid at higher concentrations (≥94.5 %) is regulated. Hydrochloric Acid: 5000 lbs (2270 kg) TQ (anhydrous) is regulated.

SARA Title III Sections 311/312 Hazardous Categories (40 CFR 370.21):

ACUTE HEALTH: Yes CHRONIC HEALTH: Yes FIRE: No REACTIVE: No PRESSURE: No

State Regulations

California Proposition 65: WARNING! This product contains a chemical (mercury) known to the state of California to cause reproductive/developmental effects.

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U.S. TSCA Inventory: Hydrochloric acid, nitric acid, and mercury (II) chloride are listed.

TSCA 12(b), Export Notification: Not listed.

Canadian Regulations: WHMIS Information is not provided for this material.

16. OTHER INFORMATION

Issue Date: 19 July 2019

Sources: ChemAdvisor, Inc., MSDS *Hydrochloric Acid*, 21 March 2014.

ChemAdvisor, Inc., MSDS Mercuric Chloride, 21 March 2014.

ChemAdvisor, Inc., MSDS Nitric Acid, 21 March 2014.

Hazardous Substances Data Bank, National Library of Medicine, *Hydrochloric Acid CAS* 7647-01-0, Animal Toxicity Studies, available at https://toxnet.nlm.nih.gov/cgi-bin/sis/htmlgen?HSDB (accessed Jul 2019).

NIOSH Pocket Guide to Chemical Hazards, *Hydrochloric Acid CAS No.* 7647-01-0, available at http://www.cdc.gov/niosh/npg/npgd0332.html (accessed Jul 2019).

Hazardous Substances Data Bank (HSDB), National Library of Medicine's TOXNET system, *Nitric Acid CAS No. 7697-37-2*; available at https://toxnet.nlm.nih.gov (accessed Jul 2019).

Key of Acronyms:

ACGIH	American Conference of Governmental Industrial	NRC	Nuclear Regulatory Commission
	Hygienists		
ALI	Annual Limit on Intake	NTP	National Toxicology Program
CAS	Chemical Abstracts Service	OSHA	Occupational Safety and Health Administration
CERCLA	Comprehensive Environmental Response,	PEL	Permissible Exposure Limit
	Compensation, and Liability Act		
CFR	Code of Federal Regulations	RCRA	Resource Conservation and Recovery Act
DOT	Department of Transportation	REL	Recommended Exposure Limit
EC50	Effective Concentration, 50 %	RM	Reference Material
EINECS	European Inventory of Existing Commercial Chemical	RQ	Reportable Quantity
	Substances		•
EPCRA	Emergency Planning and Community Right-to-Know	RTECS	Registry of Toxic Effects of Chemical Substances
	Act		
IARC	International Agency for Research on Cancer	SARA	Superfund Amendments and Reauthorization Act
IATA	International Air Transport Association	SCBA	Self-Contained Breathing Apparatus
IDLH	Immediately Dangerous to Life and Health	SRM	Standard Reference Material
LC50	Lethal Concentration, 50 %	STEL	Short Term Exposure Limit
LD50	Lethal Dose, 50 %	TLV	Threshold Limit Value
LEL	Lower Explosive Limit	TPQ	Threshold Planning Quantity
MSDS	Material Safety Data Sheet	TSCA	Toxic Substances Control Act
NFPA	National Fire Protection Association	TWA	Time Weighted Average
NIOSH	National Institute for Occupational Safety and Health	UEL	Upper Explosive Limit
NIST	National Institute of Standards and Technology	WHMIS	Workplace Hazardous Materials Information System
n.o.s.	Not Otherwise Specified		

Disclaimer: Physical and chemical data contained in this SDS are provided only for use in assessing the hazardous nature of the material. The SDS was prepared carefully, using current references; however, NIST does not certify the data in the SDS. The certified values for this material are given in the NIST Certificate of Analysis.

Users of this SRM should ensure that the SDS in their possession is current. This can be accomplished by contacting the SRM Program: telephone (301) 975-2200; fax (301) 948-3730; e-mail srmmsds@nist.gov; or via the Internet at https://www.nist.gov/srm.

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